

# Biologia evolutiva

## Origem da variabilidade genética

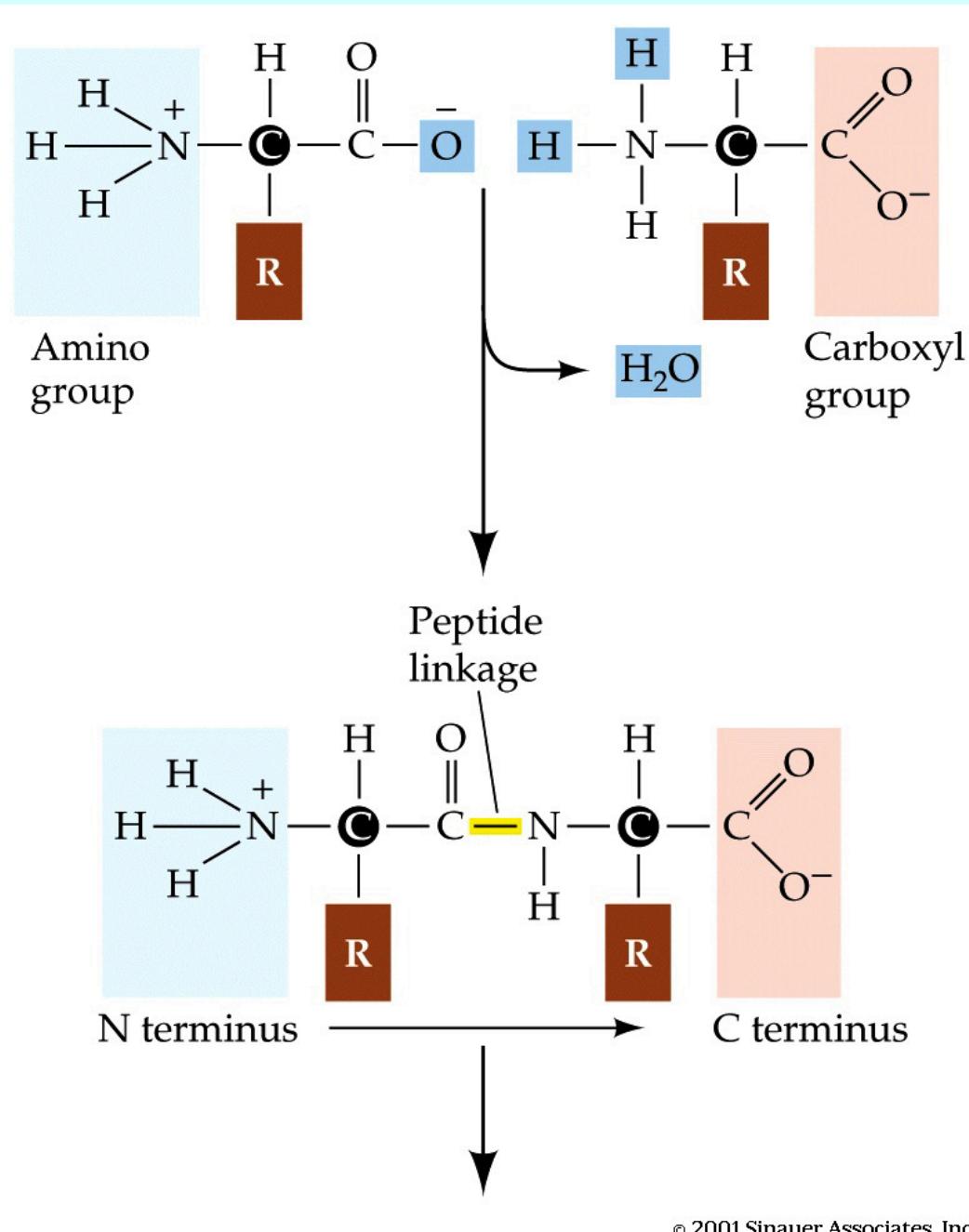
Sergio Russo Matioli

Departamento de Genética e Biologia Evolutiva

Instituto de Biociências

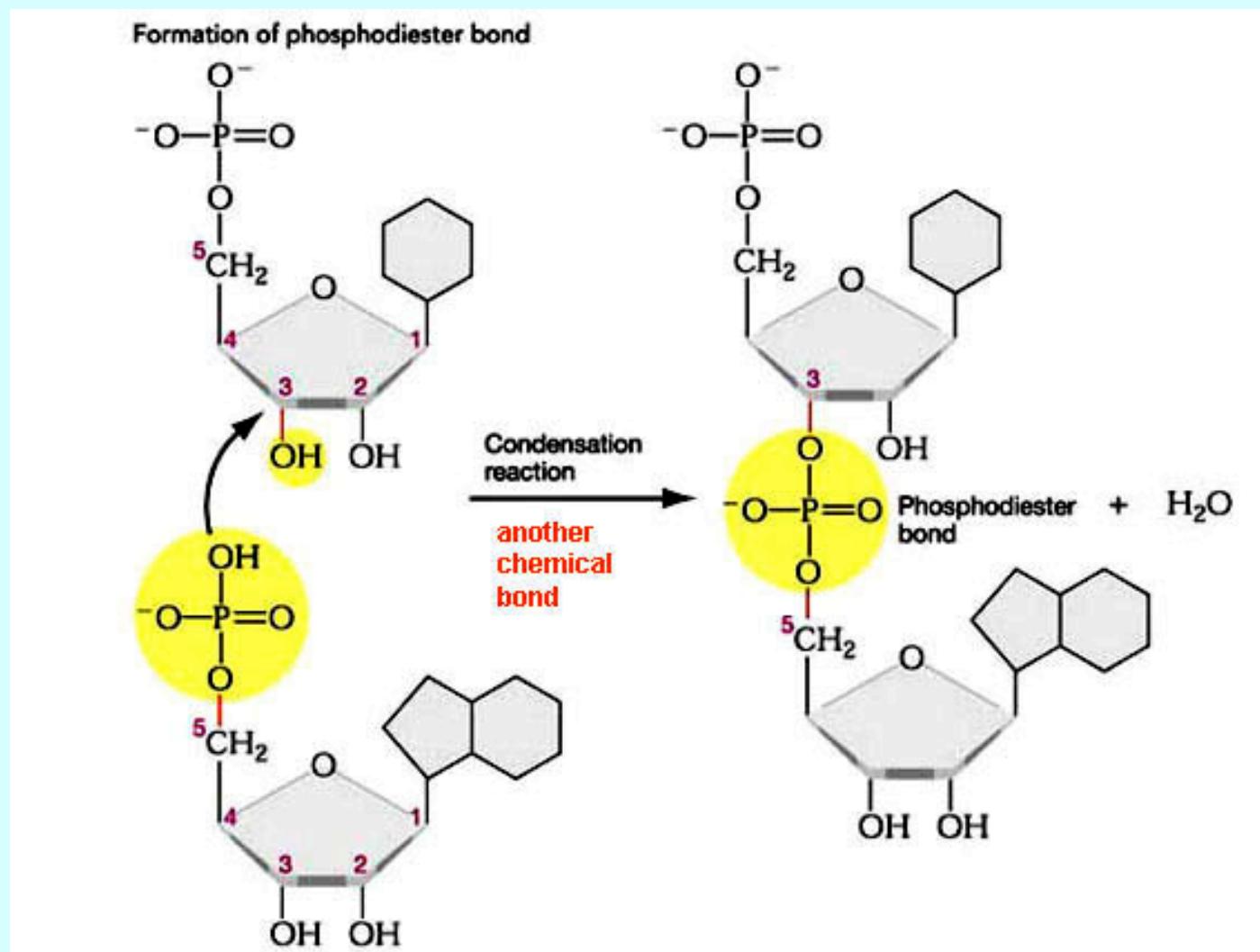
Universidade de São Paulo

# Conceitos de Bioquímica básica



Ligaçāo peptídica

# Conceitos de Bioquímica básica



Ligaçāo fosfodiéster

# Friedrich Miescher (1844-1895)



Biólogo suíço, isolou, em 1869, vários compostos químicos ricos em fósforo de núcleos de células brancas do sangue (obtidas de pus), que ele denominou como nucleína.

# Theodor Boveri (1862-1915)



Biólogo alemão, propôs, pouco após a redescoberta das leis de Mendel, conjuntamente com Walter Sutton, a teoria cromossômica da herança, na qual os fatores responsáveis pela hereditariedade estariam localizados nos cromossomos.

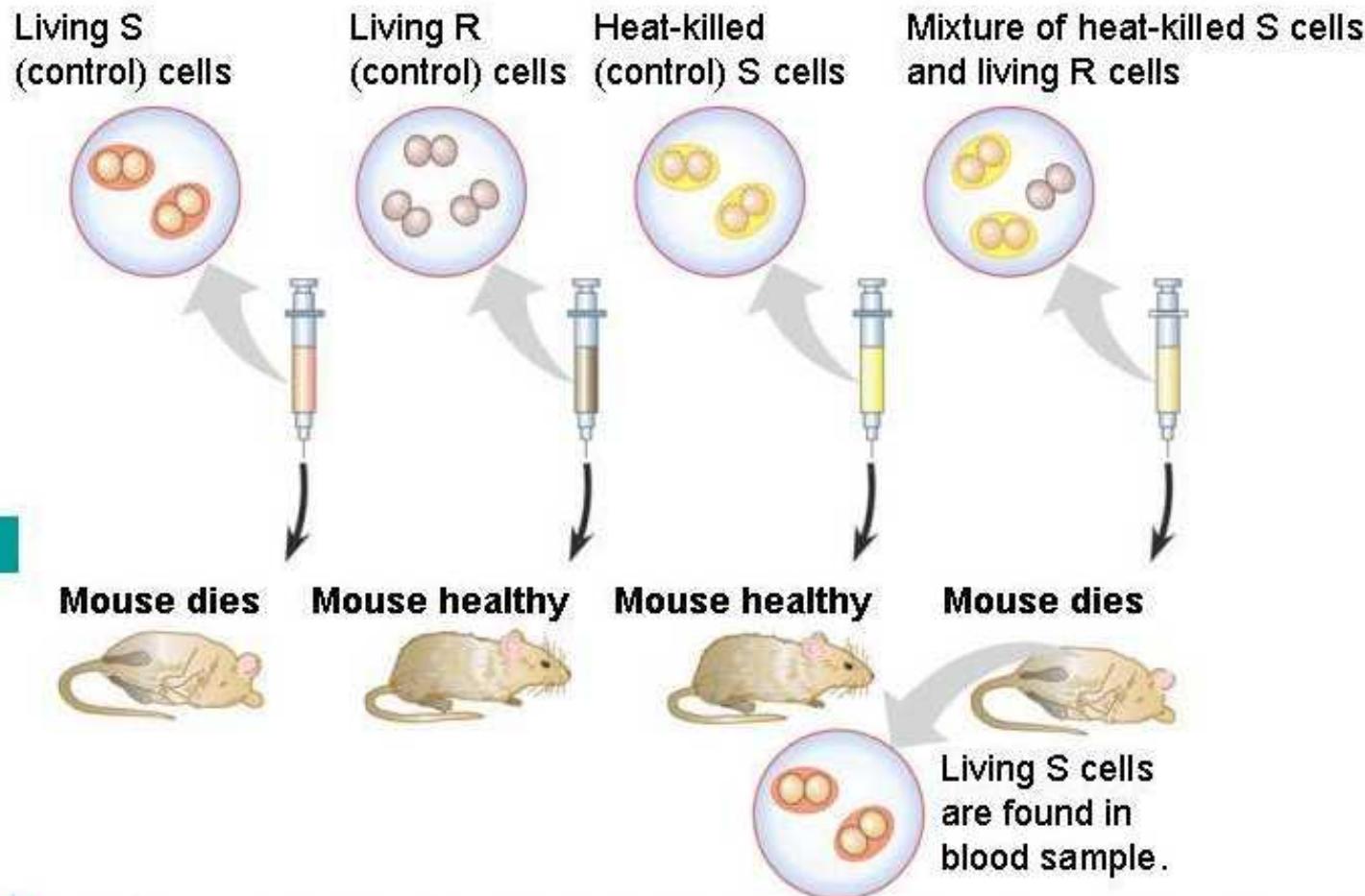
# Frederick Griffith (1879-1928)



Médico inglês, ficou conhecido por caracterizar, em 1928 o “princípio transformante”, mostrando que o material genético constituía-se em uma substância química encontrada nas células mesmo que não vivas.

# O experimento de Frederick Griffith

**EXPERIMENT** Bacteria of the “S” (smooth) strain of *Streptococcus pneumoniae* are pathogenic because they have a capsule that protects them from an animal’s defense system. Bacteria of the “R” (rough) strain lack a capsule and are nonpathogenic. Frederick Griffith injected mice with the two strains as shown below:



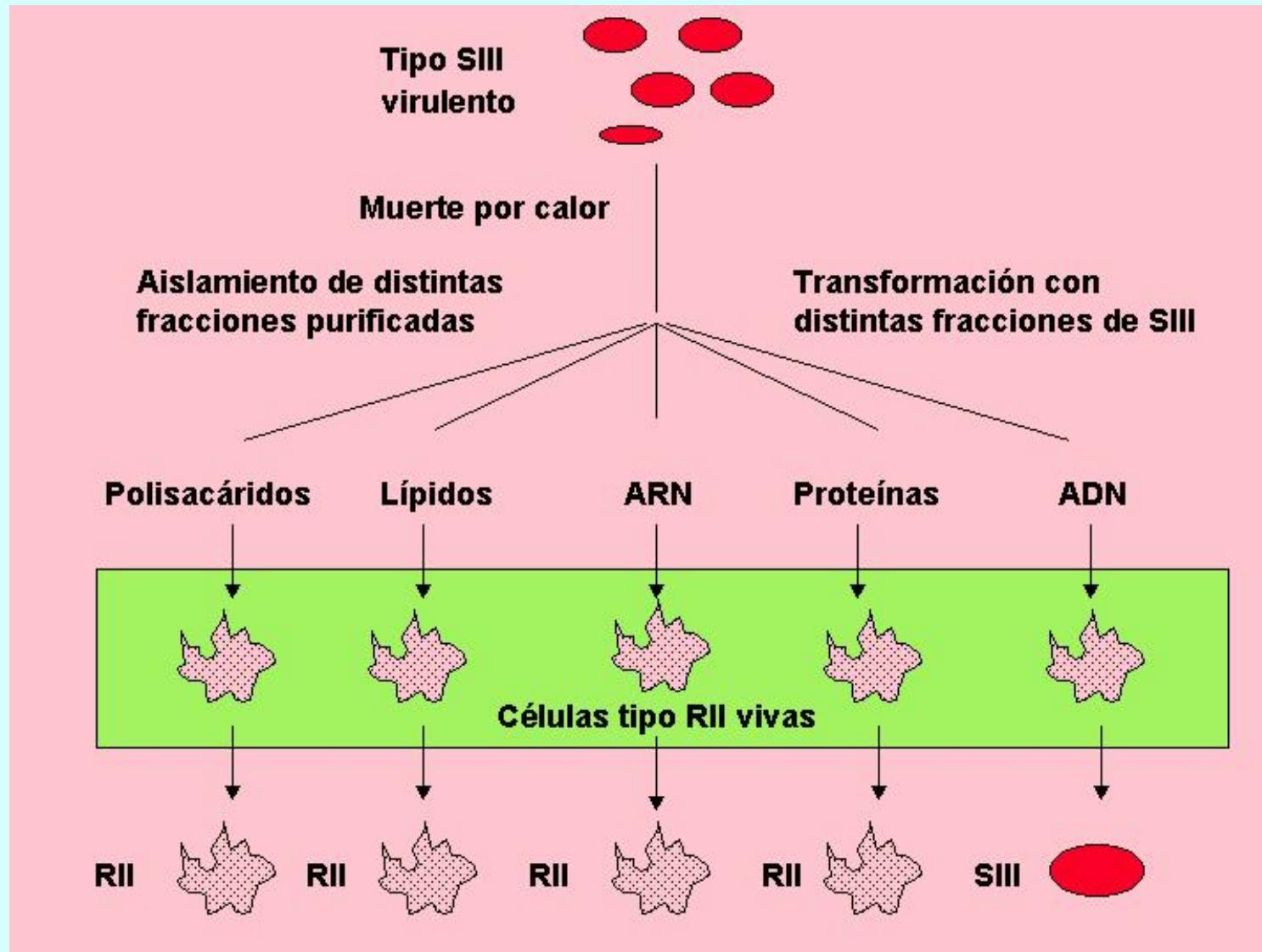
**CONCLUSION** Griffith concluded that the living R bacteria had been transformed into pathogenic S bacteria by an unknown, heritable substance from the dead S cells.

# Oswald Theodore Avery (1877-1955)

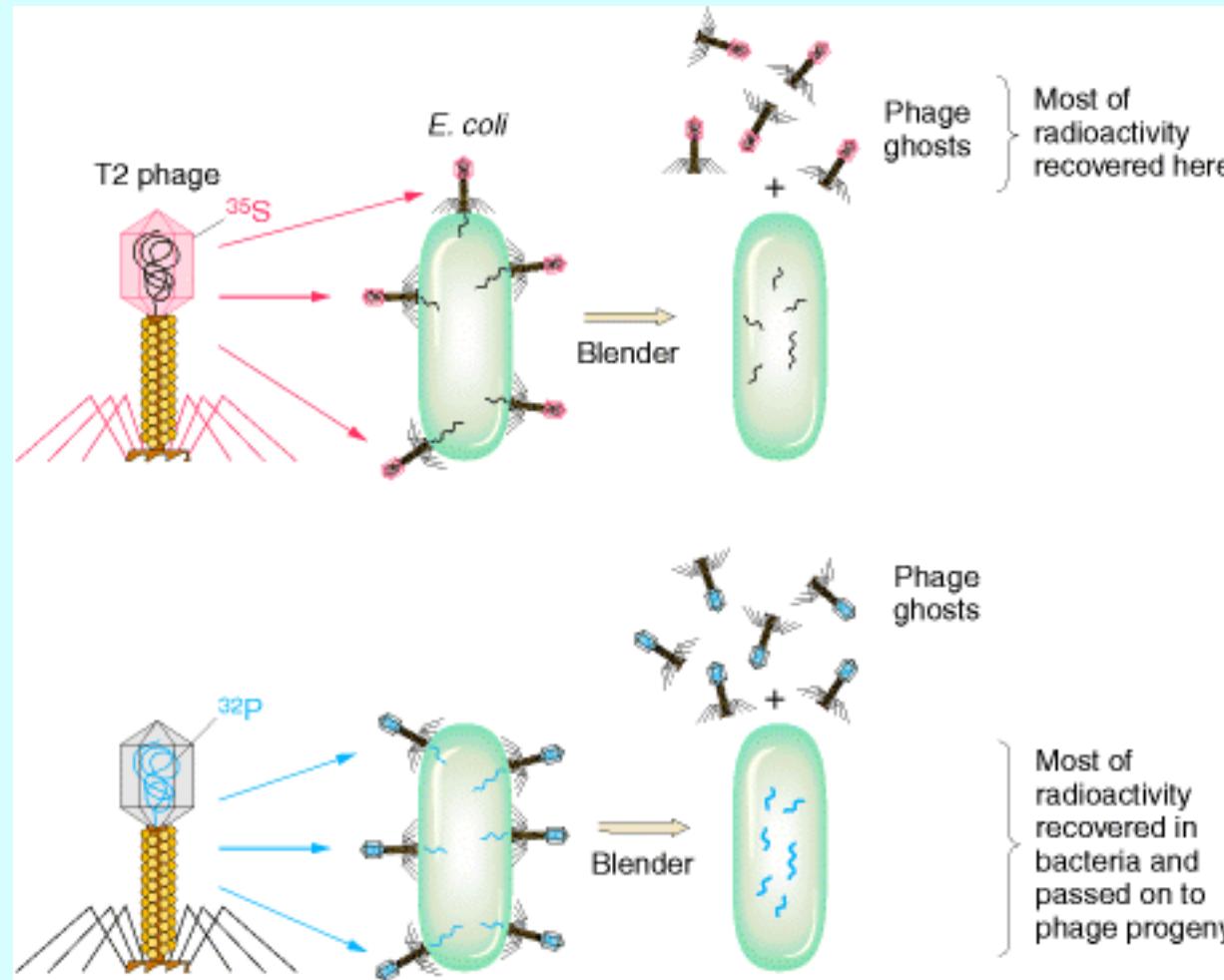


Médico canadense, juntamente com os americanos Colin MacLeod e Maclyn McCarty, demonstrou que o DNA é o material genético em 1944.

# O experimento de Avery, MacLeod e MacCarthy (1944)

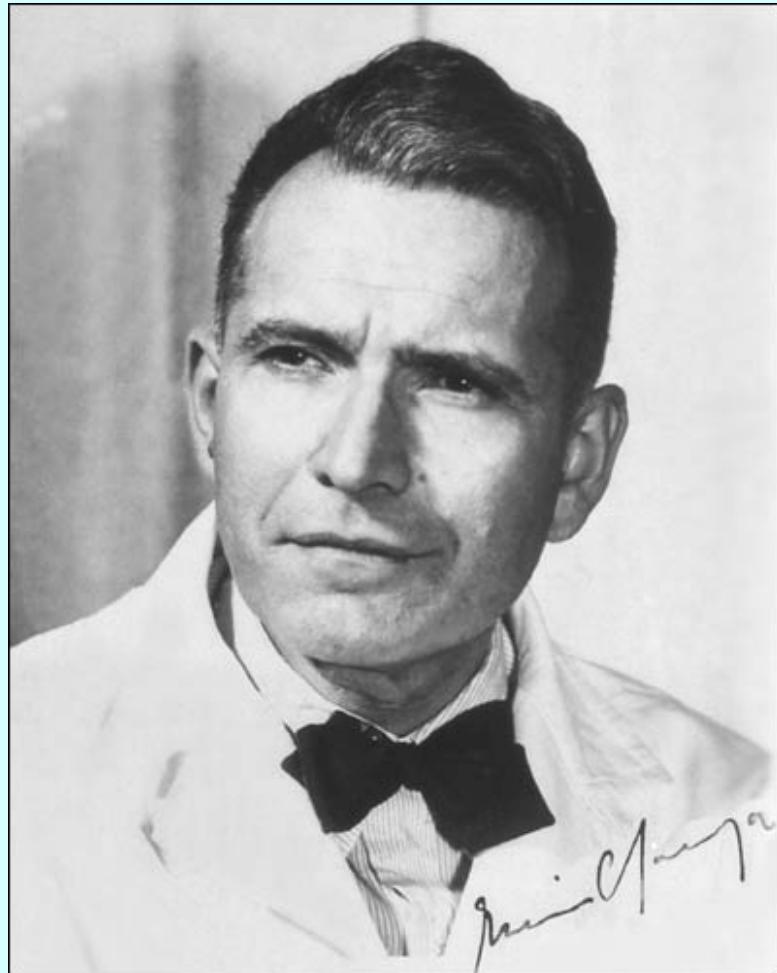


# O experimento de Hershey e Chase



**Hershey, Alfred D. and Chase, Martha (1952) Independent functions of viral protein and nucleic acid in growth of bacteriophage. J Gen Physiol. 1:39-56.**

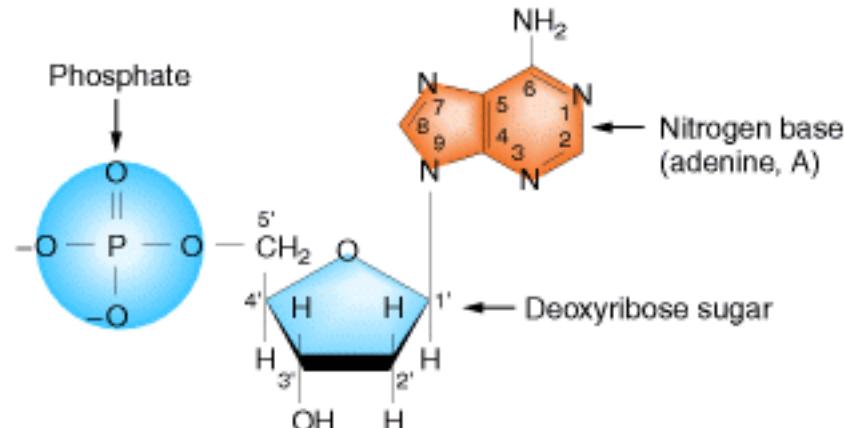
# Erwin Chargaff (1905-2002)



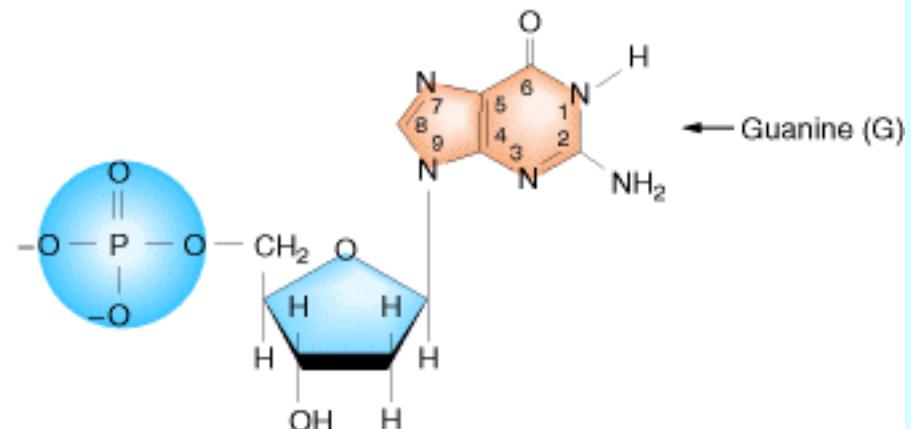
Bioquímico austríaco que emigrou para os E.U.A. durante o período nazista. Formulou as regras:  $\%A = \%T$  e  $\%C = \%G$ , importante para se desvendar a estrutura do DNA.

# Estrutura do DNA

## Purine nucleotides

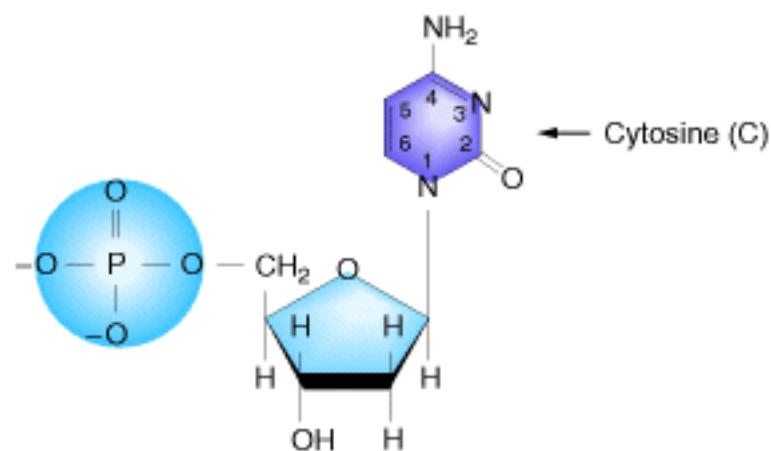


Deoxyadenosine 5'-phosphate (dAMP)

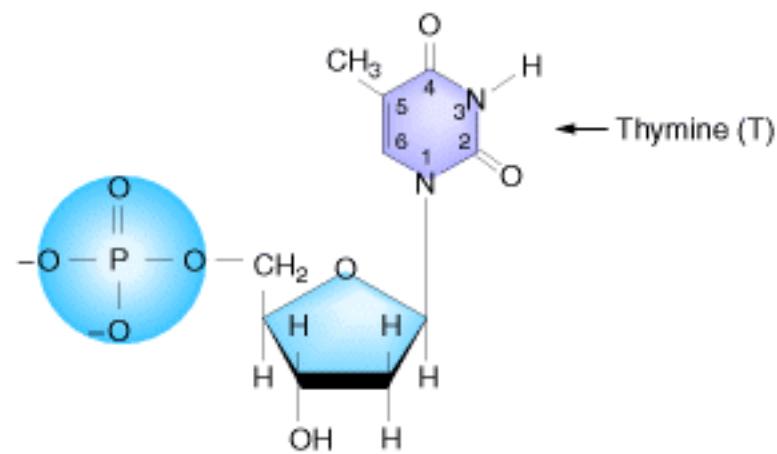


Deoxyguanosine 5'-phosphate (dGMP)

## Pyrimidine nucleotides



Deoxycytidine 5'-phosphate (dCMP)



Deoxythymidine 5'-phosphate (dTTP)

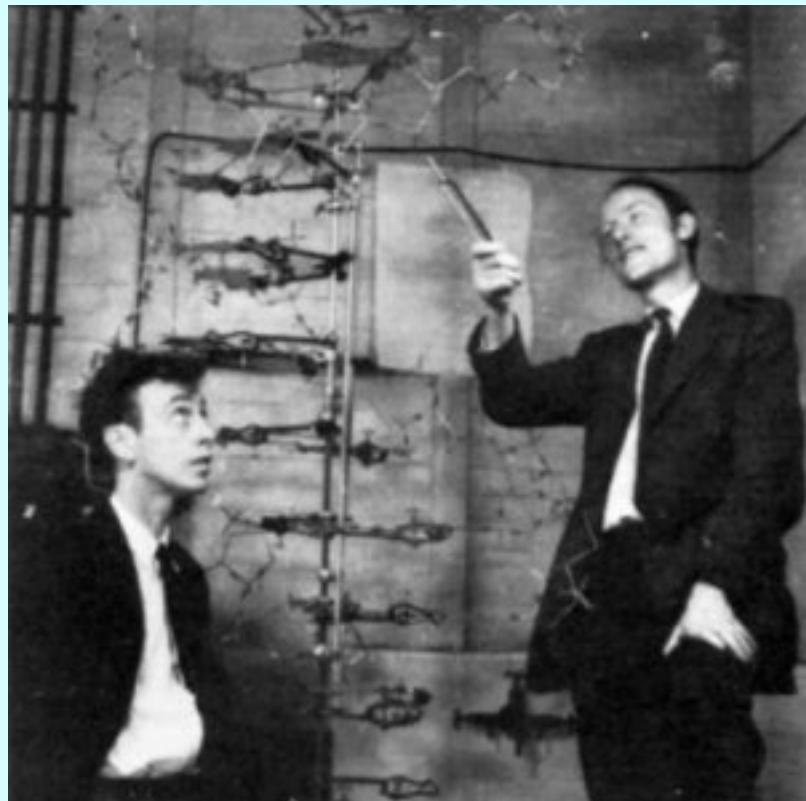
# Rosalind Franklin (1920-1958)



Biofísica inglesa, foi pioneira no uso da difração de Raios-X em biomoléculas.

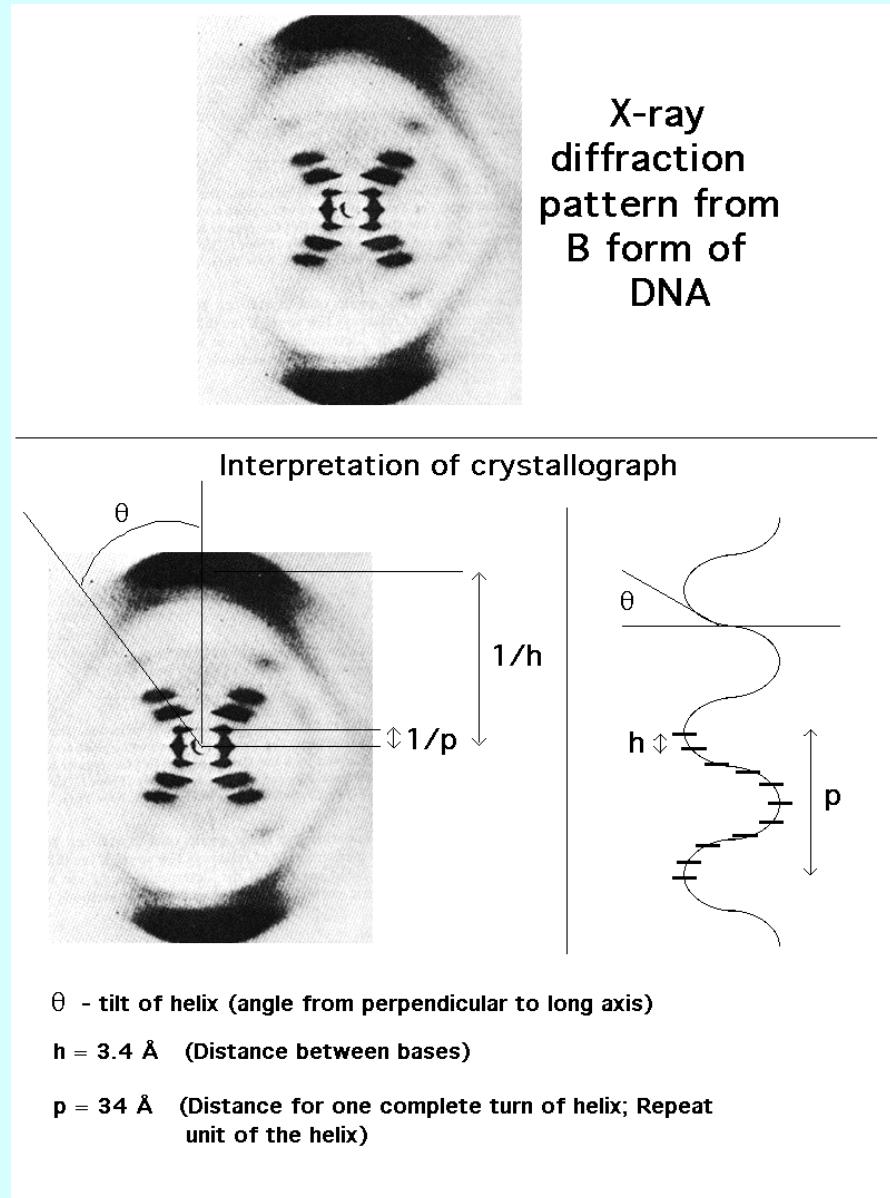
# **James Dewey Watson (1928-)**

# **Francis Harry Compton Crick (1916-2004)**



Zoólogo americano e físico inglês, propuseram um modelo de estrutura do DNA baseados em difração de Raios X de cristais da molécula obtidos por R. Franklin. Receberam o prêmio Nobel de 1963 por isso, Franklin já havia falecido.

# Interpretação estrutural da difração de Raios X por cristais de DNA



## MOLECULAR STRUCTURE OF NUCLEIC ACIDS

### A Structure for Deoxyribose Nucleic Acid

We wish to suggest a structure for the salt of deoxyribonucleic acid (DNA). This structure has novel features which are of considerable biological interest.

A structure for nucleic acid has already been proposed by Franklin and Crick<sup>1</sup>. They kindly made their manuscript available to us in advance of publication. Their model consists of three intertwined chains, with the phosphate near the fibre axis, and the bases on the outside. In our opinion, this structure is unsatisfactory for two reasons: (1) We believe that the material which gives the X-ray diagram is the salt, not the free acid. Without the acidic hydroxyl groups it is not clear what forces would hold the structure together, especially as the negatively charged phosphates near the axis will repel each other. (2) Some of the van der Waals distances appear to be too small.

Another three-chain structure has also been suggested by Fraser (in the press). In his model the phosphates are on the outside and the bases on the inside linked together by hydrogen bonds. This structure as described is rather disordered, and for this reason we shall not comment on it.

We wish to put forward a radically different structure for the salt of deoxyribonucleic acid. This structure has two helical chains each coiled round the same axis (see diagram). We have made the usual chemical assumptions, namely, that each chain consists of phosphate diester groups joining 2-deoxyribose residues with  $3',5'$  linkages. The two chains (but not their bases) are related by a dyad perpendicular to the fibre axis. Both chains follow right-handed helices, but owing to the dyad the sequence of the atoms in the two chains run in opposite directions. Each chain loosely resembles Faurberg's<sup>2</sup> model No. I; that is, the bases are on the inside of the helix and the phosphates on the outside. The configuration of the sugar and the atoms near it is close to Faurberg's "standard conformation", the sugar being roughly perpendicular to the attached base. There is a residue on each chain every  $2.4 \text{ \AA}$ . in the  $z$ -direction. We have assumed an angle of  $36^\circ$  between adjacent residues in the same chain, so that the structure repeats after 10 residues on each chain, that is, after  $34 \text{ \AA}$ . The distance of a phosphate atom from the fibre axis is  $10 \text{ \AA}$ . As the phosphates are on the outside, cations have easy access to them.

The structure is an open one, and its water content is rather high. At lower water contents we would expect the bases to tilt so that the structure could become more compact.

The novel feature of the structure is the manner in which the two chains are held together by the purine and pyrimidine bases. The planes of the bases are perpendicular to the fibre axis. They are joined together in pairs, a single base from one chain being hydrogen-bonded to a single base from the other chain, so that the two lie side by side with identical  $z$ -coordinates. One of the pair must be a purine and the other a pyrimidine for bonding to occur. The hydrogen bonds are made as follows: purine position 1 to pyrimidine position 1; purine position 6 to pyrimidine position 6.

If it is assumed that the bases only occur in the structure in the most plausible tautomeric forms (that is, with the keto rather than the enol configurations), it is found that only specific pairs of bases can bond together. These pairs are: adenine (purine) with thymine (pyrimidine), and guanine (purine) with cytosine (pyrimidine).

In other words, if an adenine forms one member of a pair on either chain, then on these assumptions the other member must be thymine; similarly for



This figure is purely diagrammatic. It can represent the two phosphate-diester chains, and the horizontal axis the pair of bases which are hydrogen-bonded together. The vertical line marks the fibre axis.

guanine and cytosine. The sequence of bases on a single chain does not appear to be restricted in any way. However, if only specific pairs of bases can be formed, it follows that if the sequence of bases on one chain is given, then the sequence on the other chain is automatically determined.

It has been found experimentally<sup>3,4</sup> that the ratio of the amounts of adenine to thymine, and the ratio of guanine to cytosine, are always very close to unity for deoxyribonucleic acid.

It is probably impossible to build this structure with a ribose sugar in place of the deoxyribose, as the extra oxygen atom would make too close a van der Waals contact.

The previously published X-ray data<sup>5,6</sup> on deoxyribonucleic acid are insufficient for a rigorous test of our structure. So far as we can tell, it is roughly compatible with the experimental data, but it must be regarded as unproved until it has been checked against more exact results. Some of these are given in the following communications. We were not aware of the details of the results presented there when we devised our structure, which rests mainly though not entirely on published experimental data and stereochemical arguments.

It has not escaped our notice that the specific pairing we have postulated immediately suggests a possible copying mechanism for the genetic material.

Full details of the structure, including the conditions assumed in building it, together with a set of co-ordinates for the atoms, will be published elsewhere.

We are much indebted to Dr. Jerry Donohue for constant advice and criticism, especially on interatomic distances. We have also been stimulated by a knowledge of the general nature of the unpublished experimental results and ideas of Dr. M. H. F. Wilkins, Dr. R. E. Franklin and their co-workers at King's College, London. One of us (J. D. W.) has been aided by a fellowship from the National Foundation for Infantile Paralysis.

J. D. WATSON  
F. H. C. CRICK  
Medical Research Council Unit for the Study of the Molecular Structure of Biological Systems,  
Cavendish Laboratory, Cambridge.  
April 2.

<sup>1</sup>Franklin, L., and Gosling, R. E. *Nature*, **193**, 782 (1951); *Proc. Roy. Soc. (B)*, **207**, 61 (1950).

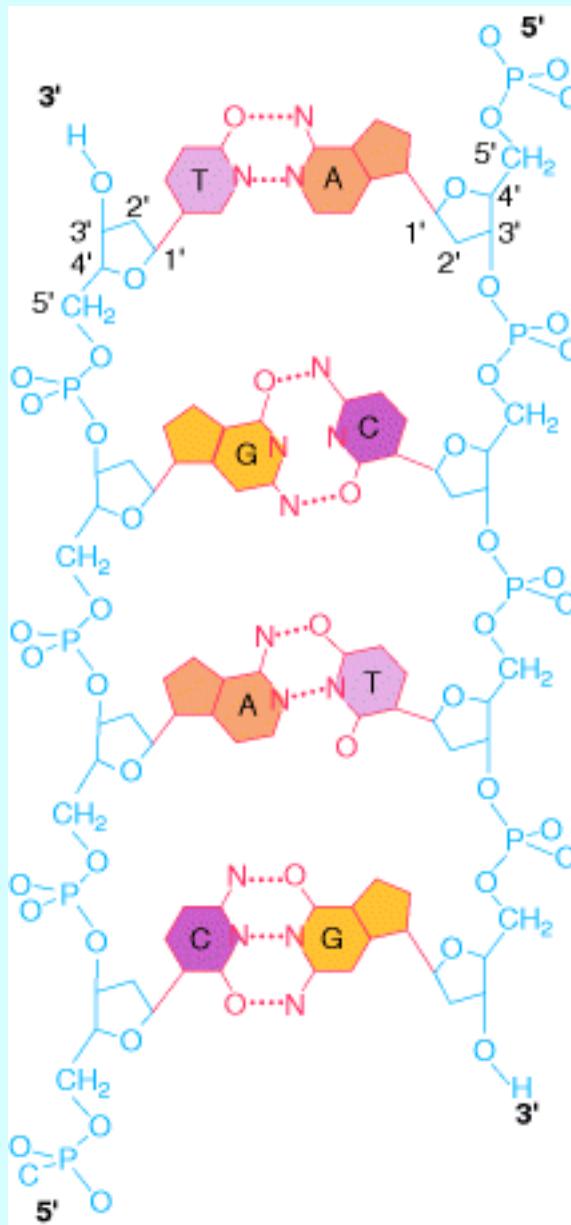
<sup>2</sup>Faurberg, S., *Acta Chem. Scand.*, **8**, 654 (1954).

<sup>3</sup>Chargaff, E., *for references see Donohue, J., Wilkins, M. H. F., and Fankhauser, J. *Science*, **106**, 166 (1947); *Journal of Molecular Biology*, **3**, 161 (1951).*

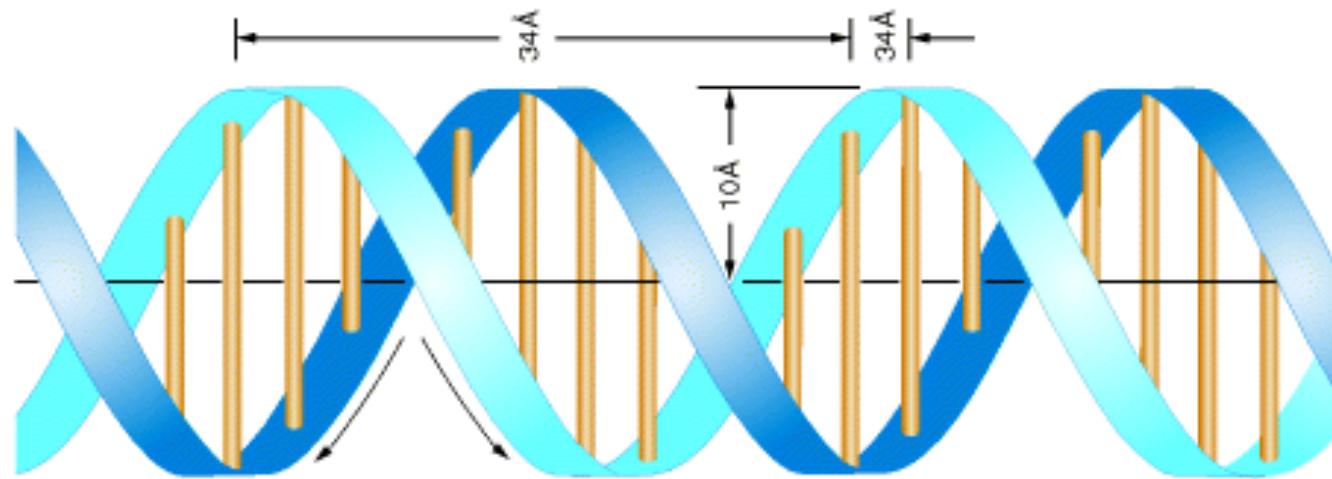
<sup>4</sup>Watson, J. D., and Crick, F. H. C. *Nature*, **195**, 730 (1962).

<sup>5</sup>Wilkins, M. H. F., and Randall, J. T., *Nucleic Acid*, **10** (Cambridge Press, 1957).

# Estrutura antiparalela do DNA

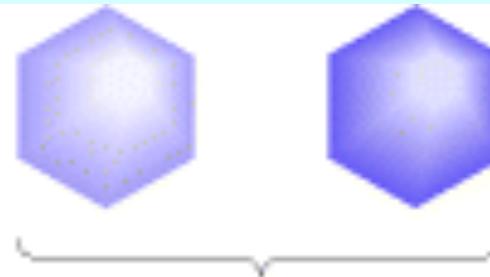


# Estrutura do DNA: dimensões

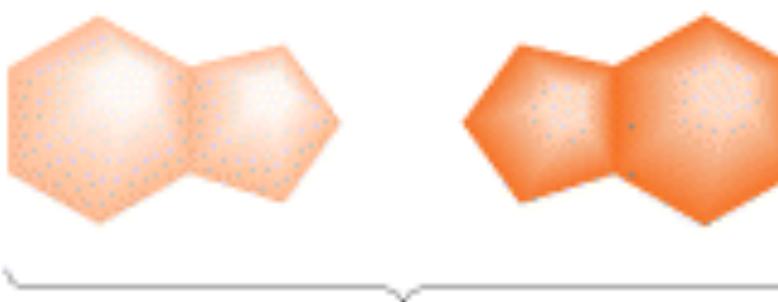


# Estrutura do DNA:bases púricas e pirimídicas

Pyrimidine + pyrimidine: DNA too thin



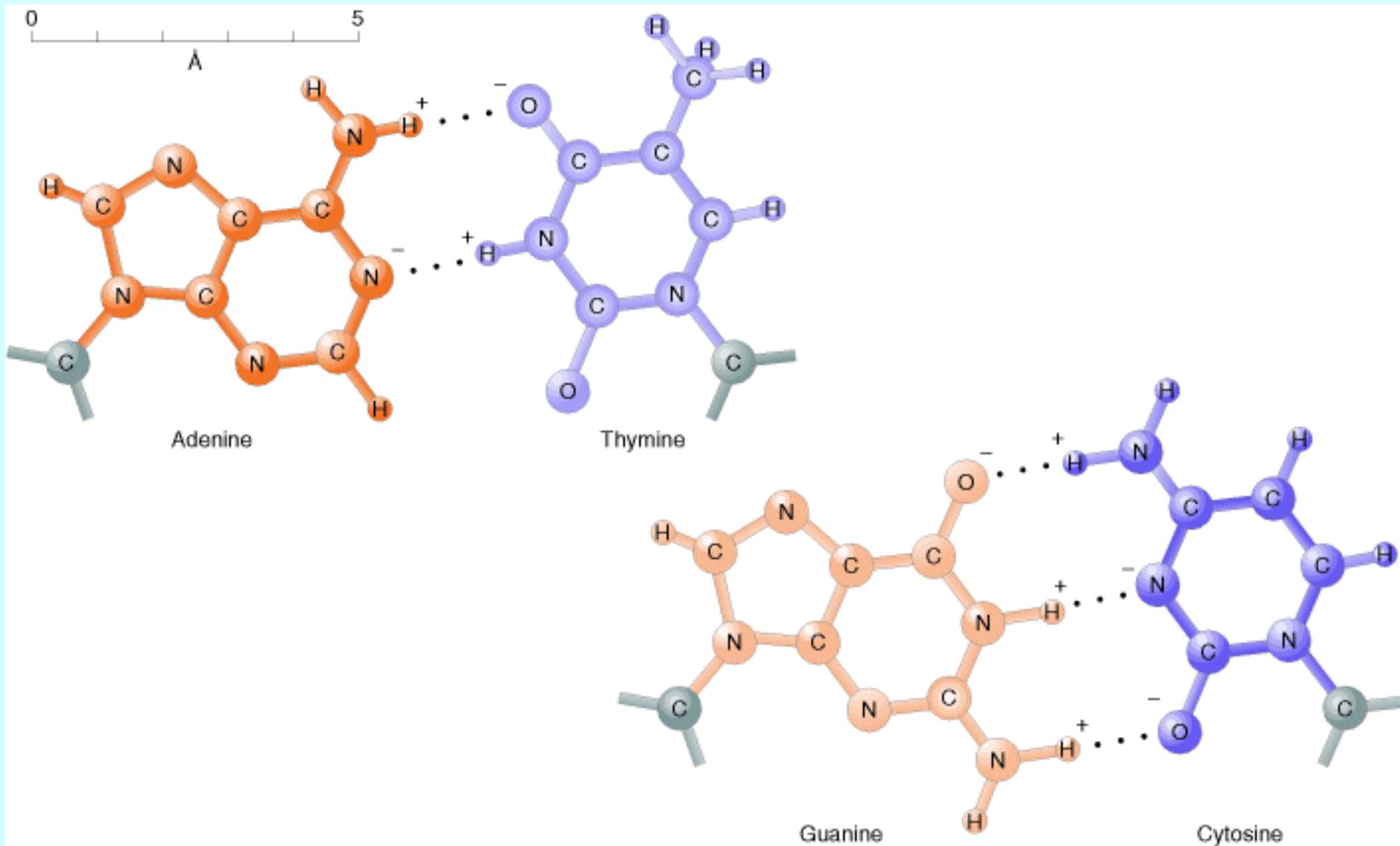
Purine + purine: DNA too thick



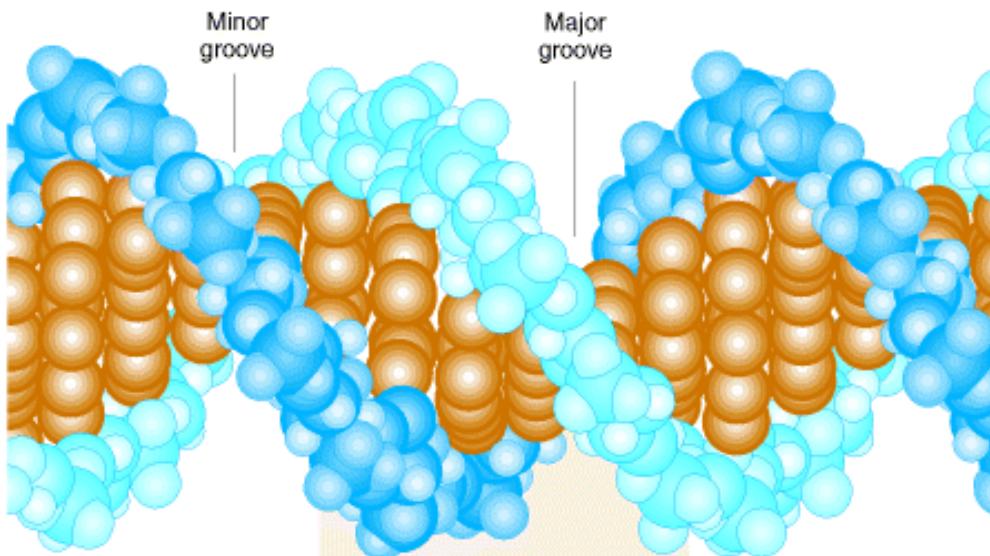
Purine + pyrimidine: thickness compatible with X-ray data



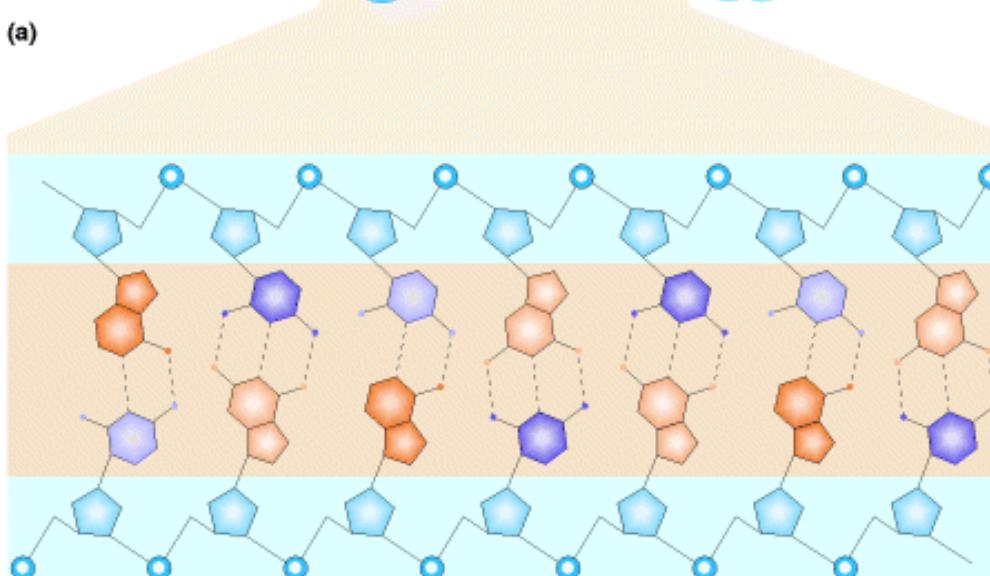
# Estrutura do DNA: pontes de hidrogênio



# Estrutura do DNA: sulcos

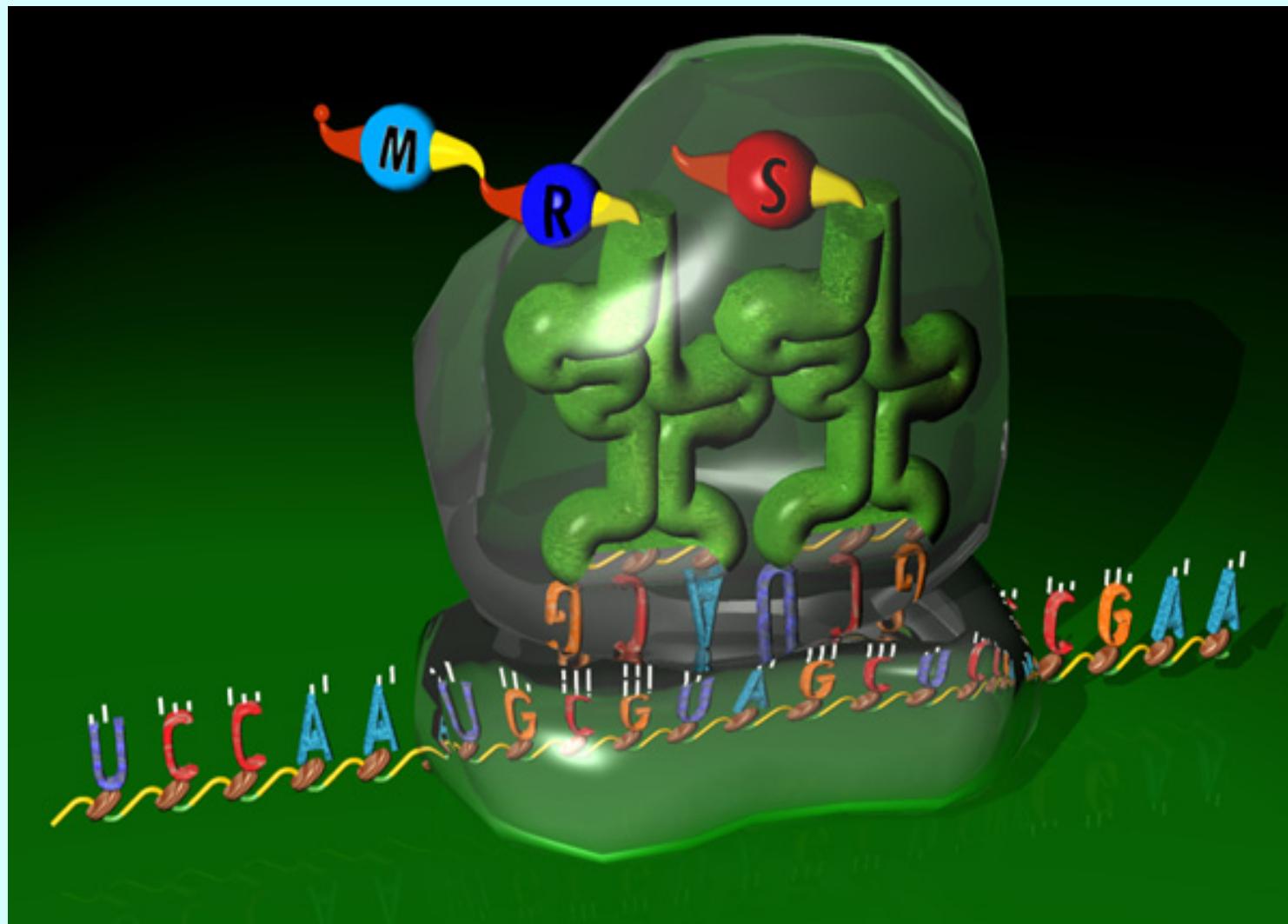


(a)

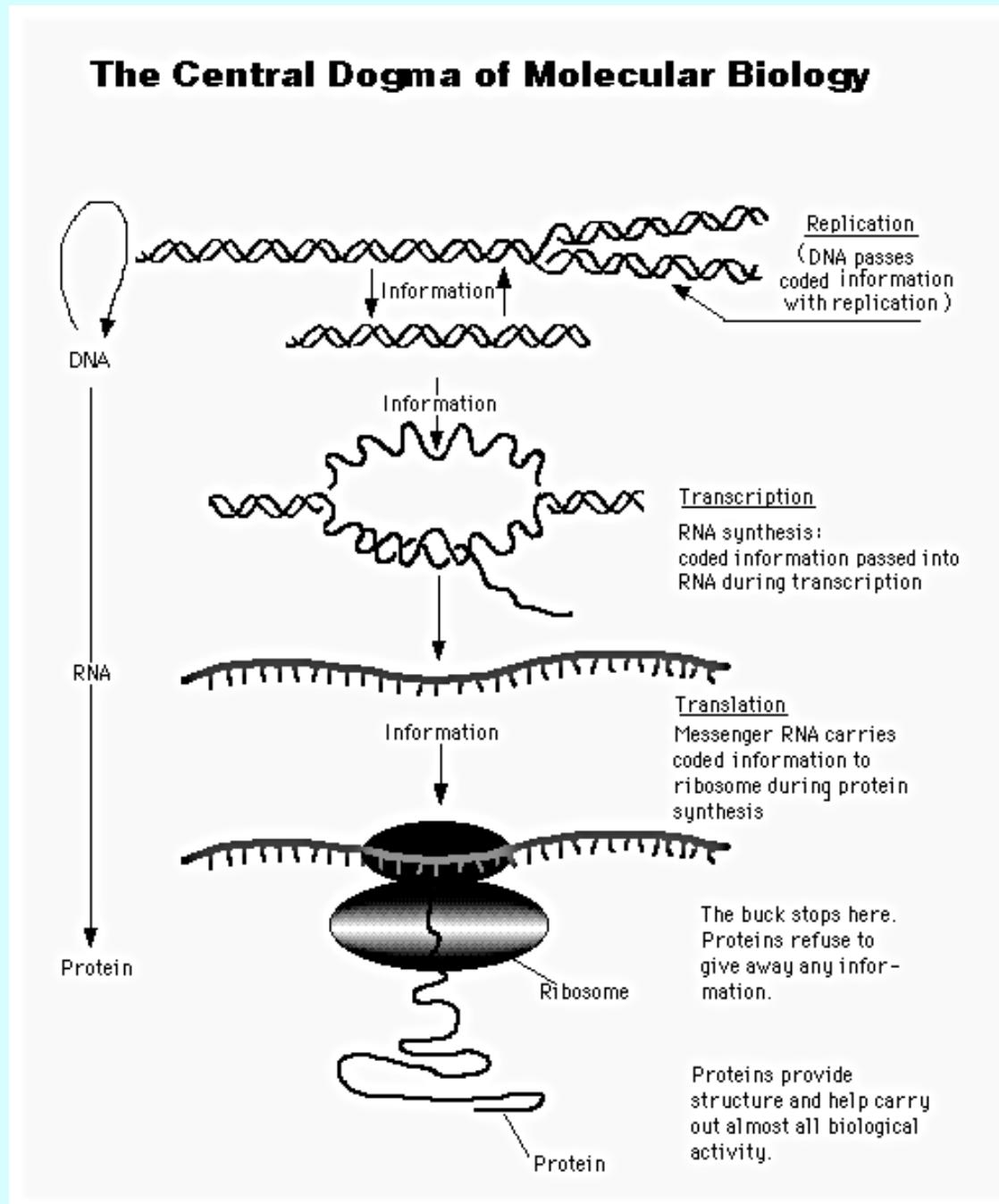


(b)

# Código genético



# O Dogma central da Biologia molecular



# Código genético

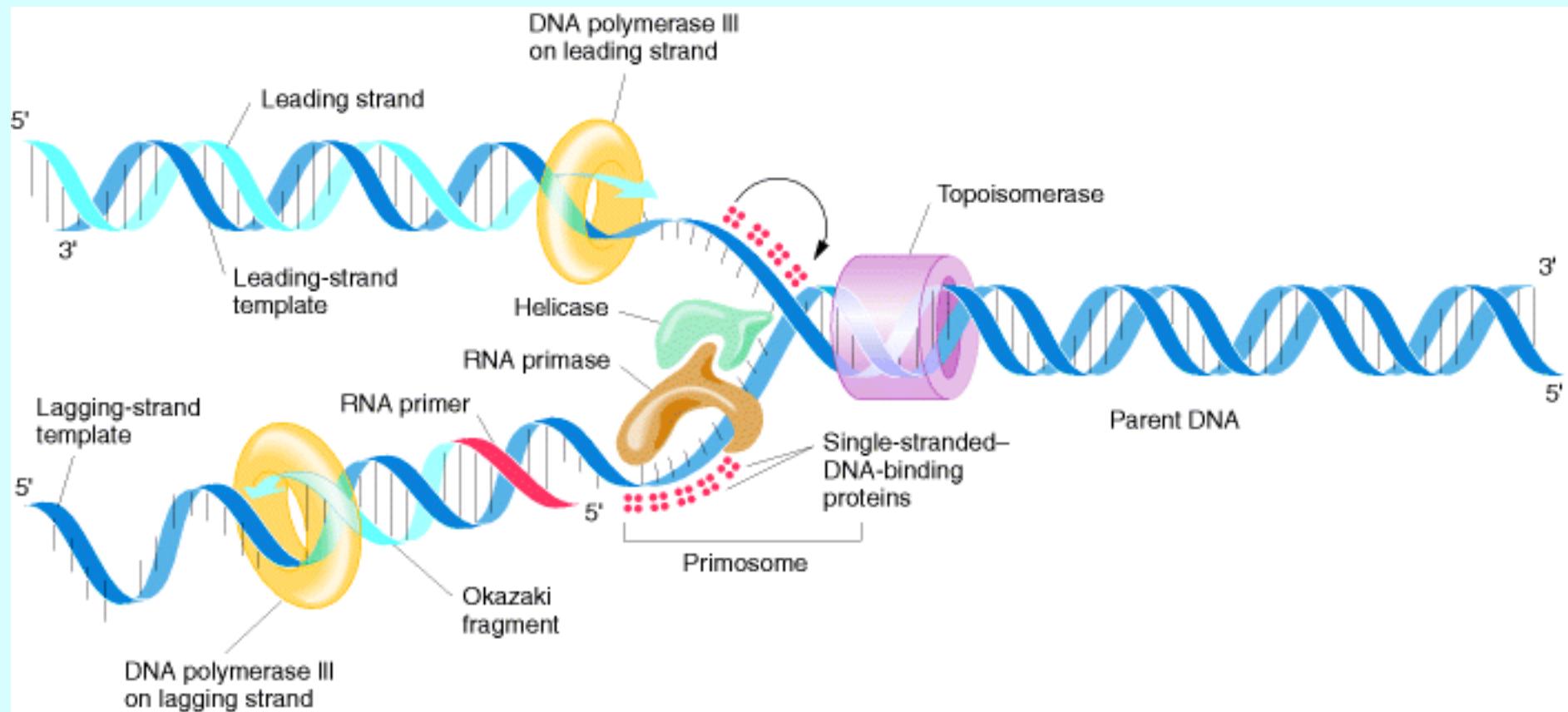
Segunda base do códon

	U	C	A	G	
U	UUU UUC UUA UUG	UCU UCC UCA UCG	UAU UAC UAA UAG	UGU UGC UGA UGG	Cys Tyr Trp
C	CUU CUC CUA CUG	CCU CCC CCA CCG	CAU CAC CAA CAG	CGU CGC CGA CGG	His Pro Gln
A	AUU AUC AUA AUG	ACU ACC ACA ACG	AAU AAC AAA AAG	AGU AGC AGA AGG	Ser Arg
G	GUU GUC GUA GUG	GCU GCC GCA GCG	GAU GAC GAA CAG	GGU GGC GGA GGG	Asp Ala Glu Gly

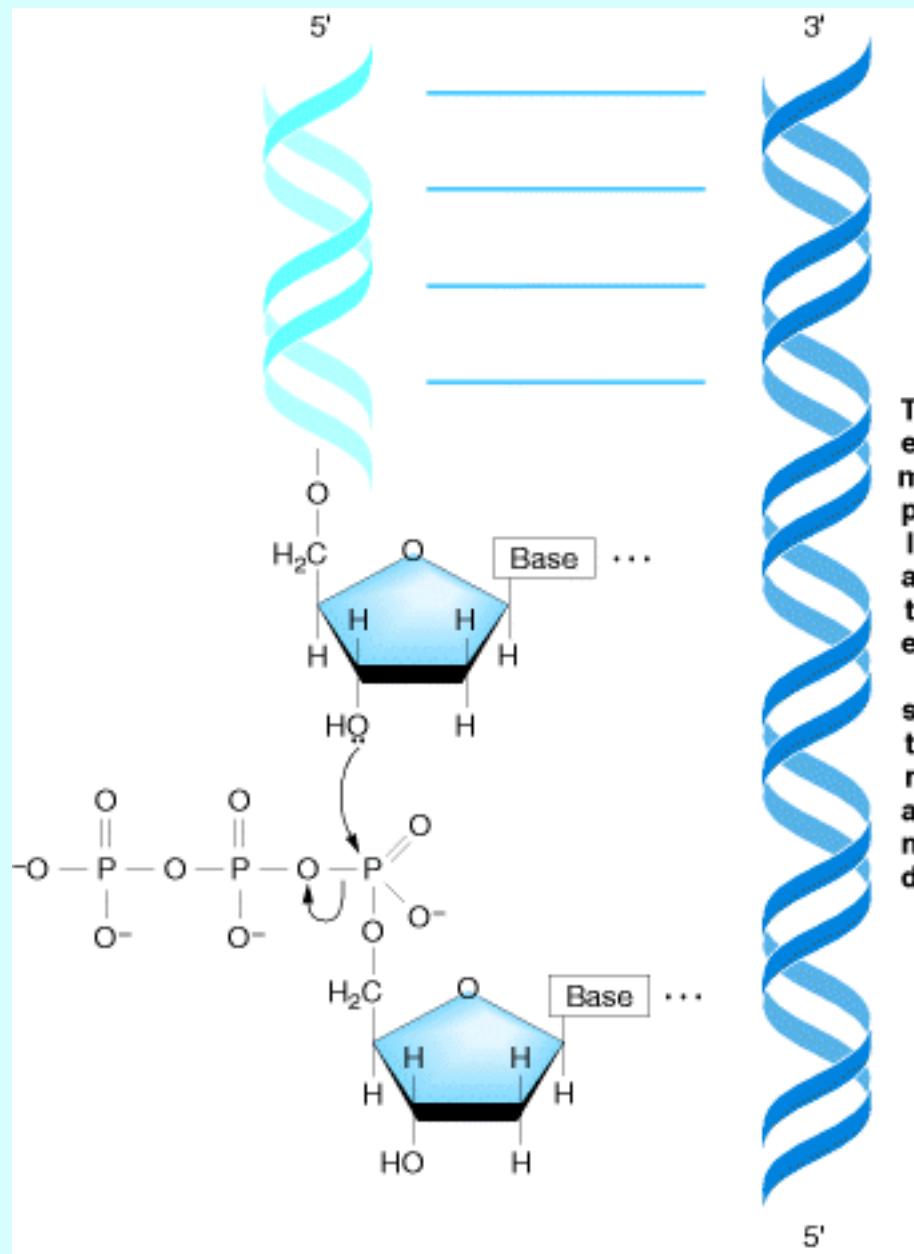
Terceira base do códon

O código genético, escrito por convenção na forma na qual os códons aparecem no mRNA. Os três códons de terminação, UAA, UAG e UGA, estão no quadro em vermelho; o códon iniciador, AUG, está mostrado em verde.

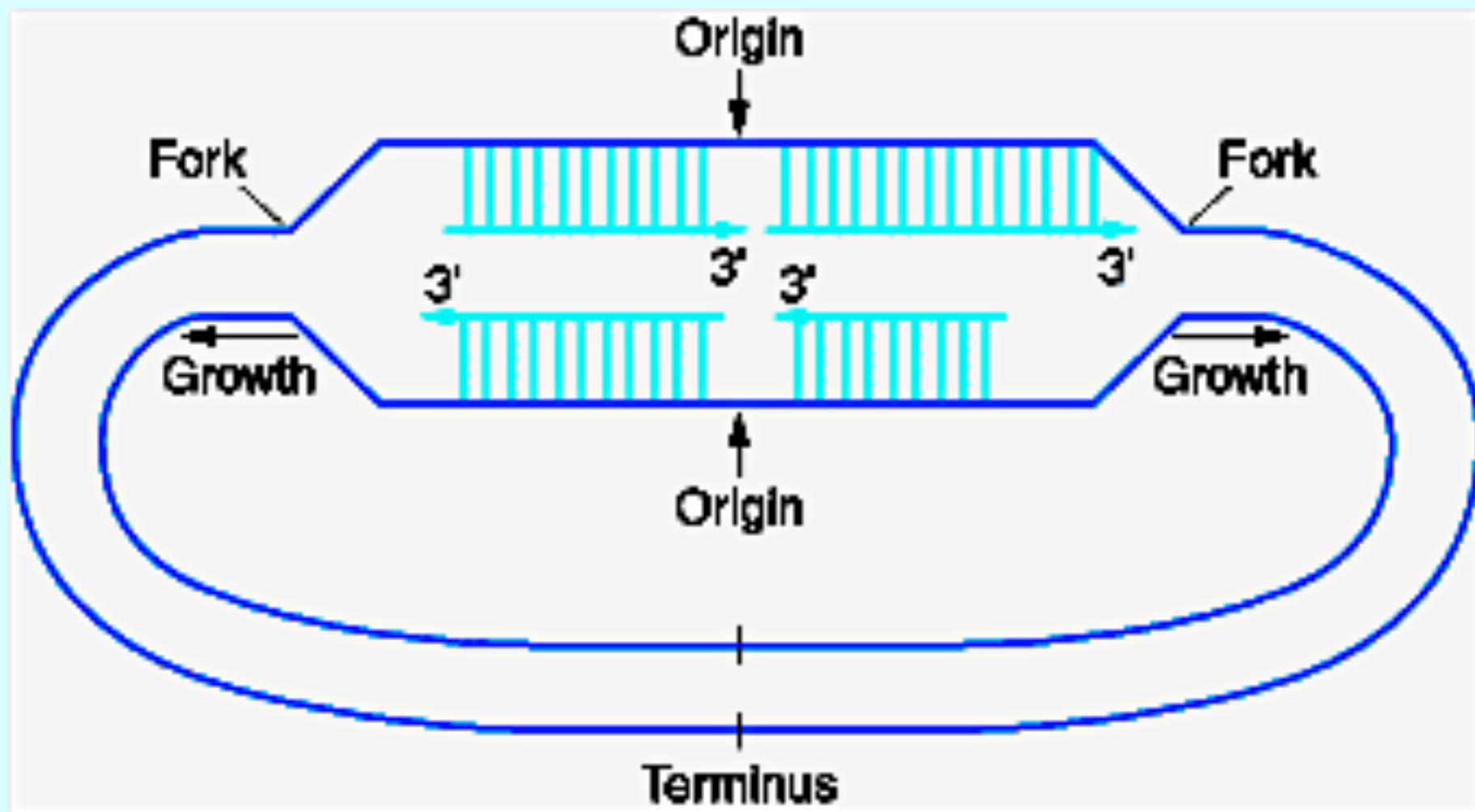
# Replicação do DNA



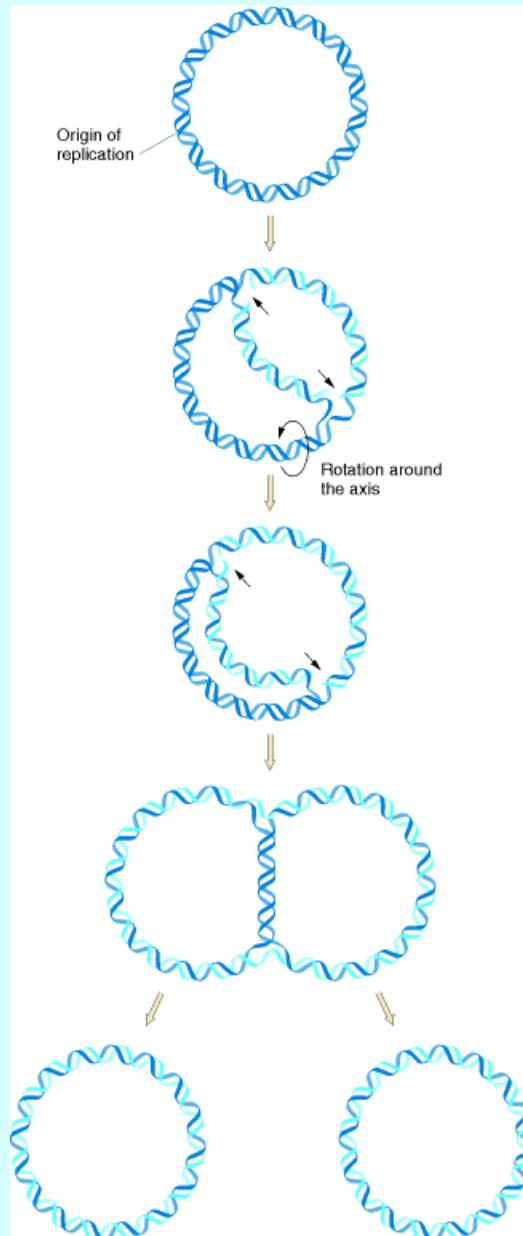
# Replicação do DNA



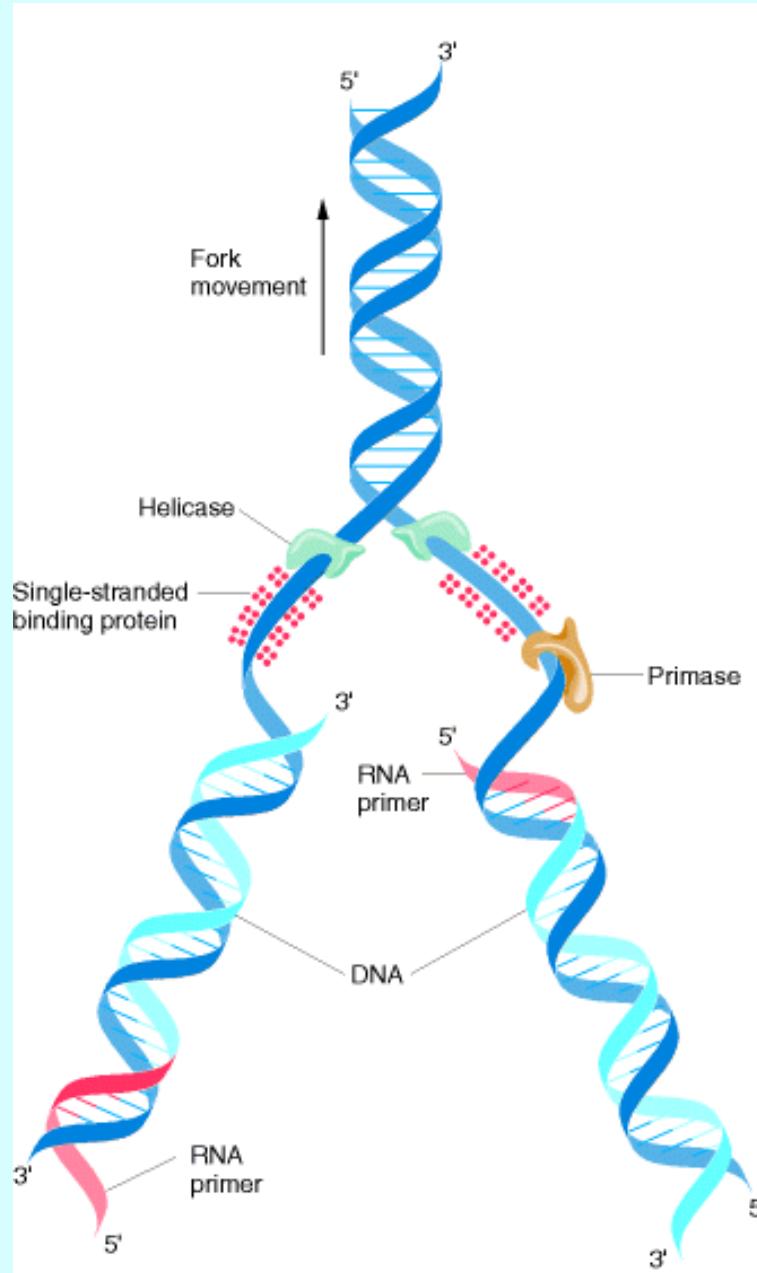
# Replicação do DNA circular



# Replicação do DNA circular



# Enzimas da replicação do DNA



# Filmes de replicação do DNA

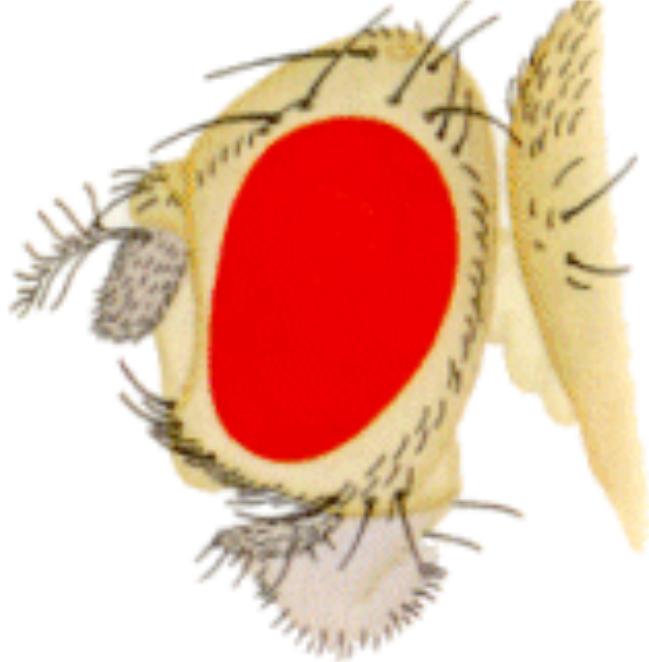
Filmes de animação mostrando a replicação do DNA

Ver em:

[https://www.youtube.com/watch?v=vNXFk\\_d6y80](https://www.youtube.com/watch?v=vNXFk_d6y80)

[http://www.dnatube.com/search/?search\\_id=dna+replication](http://www.dnatube.com/search/?search_id=dna+replication)

# Mutação e recombinação



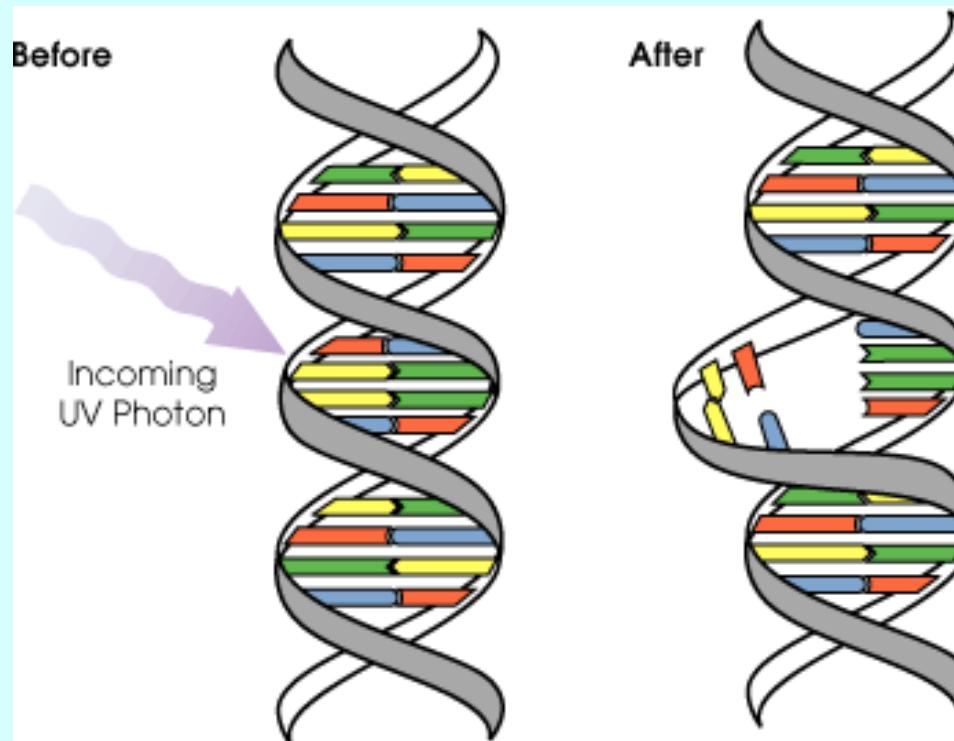
Wild-Type



White

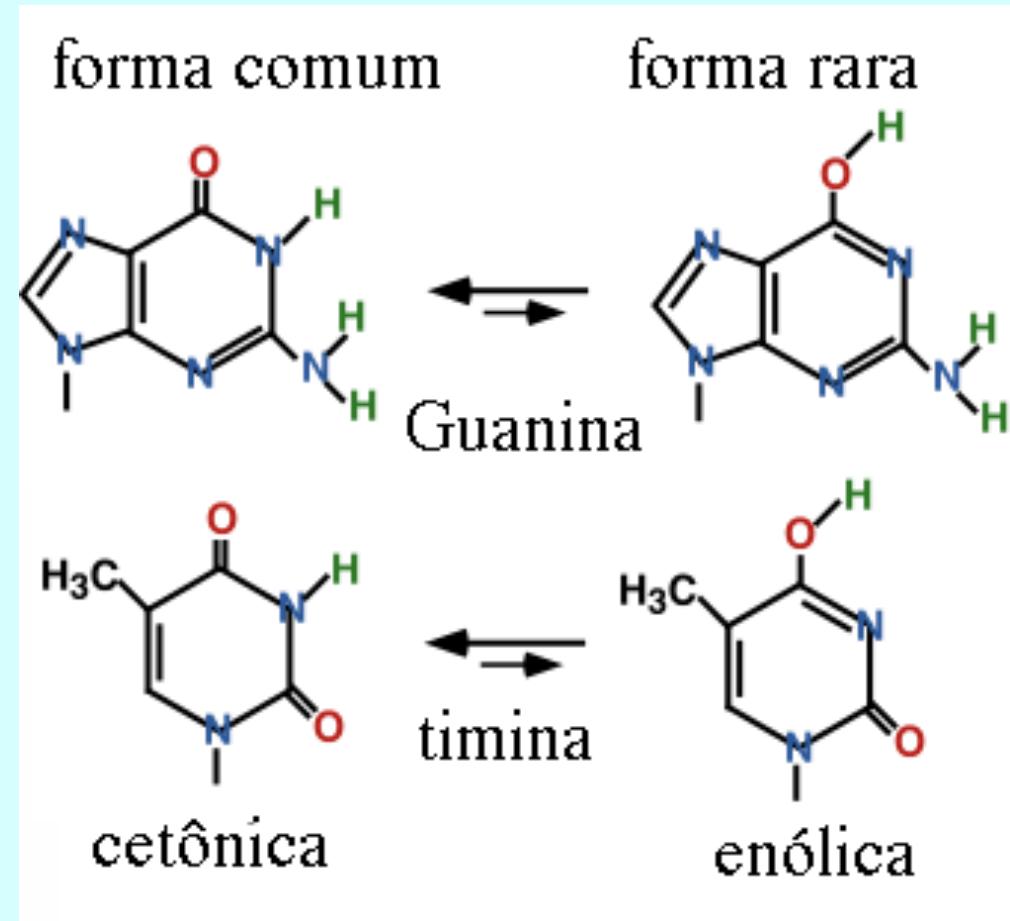
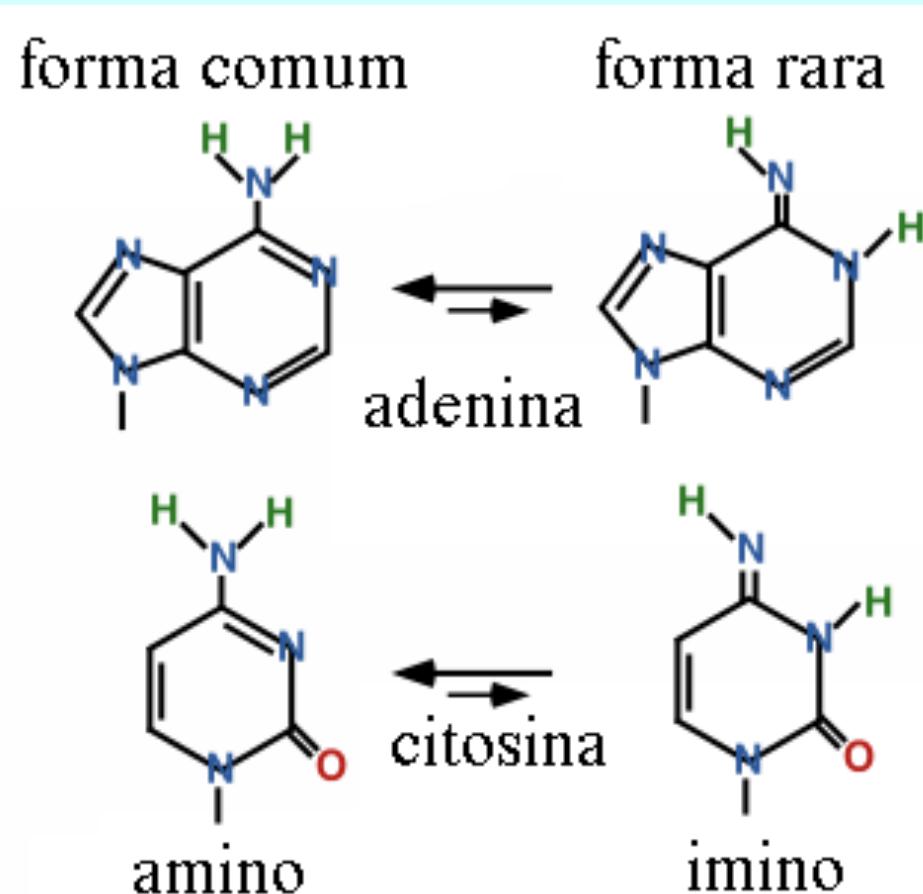
# Mutações

Hermann Müller, em 1943, recebe o prêmio Nobel por seus trabalhos com radiações ionizantes provocando mutações gênicas



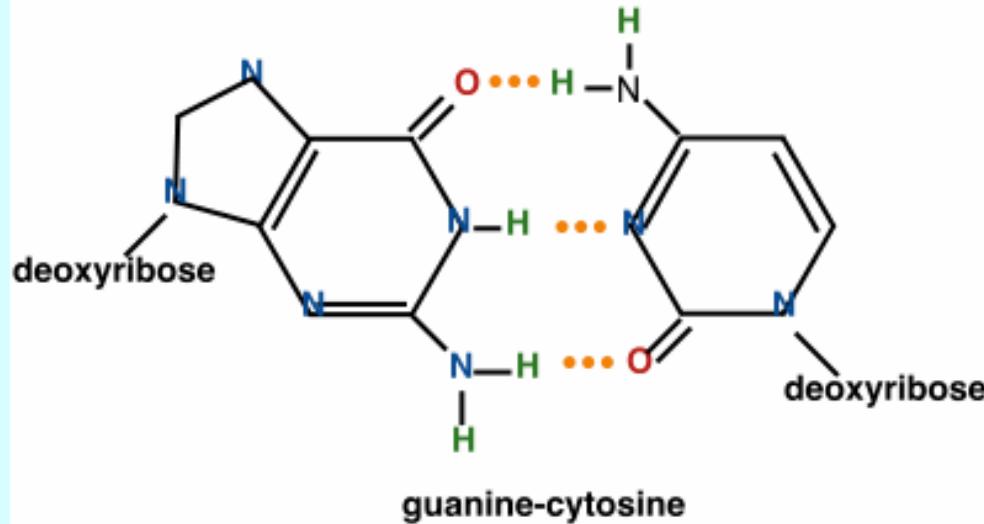
# Mutações espontâneas

# Tautômeros de bases nitrogenadas



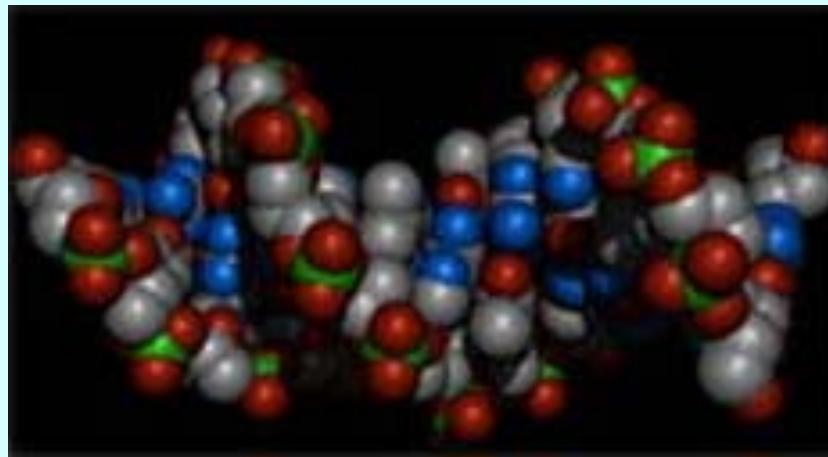
# Mutações espontâneas

## Tautomer Mispairing



# Mutações espontâneas

## Radicais livres



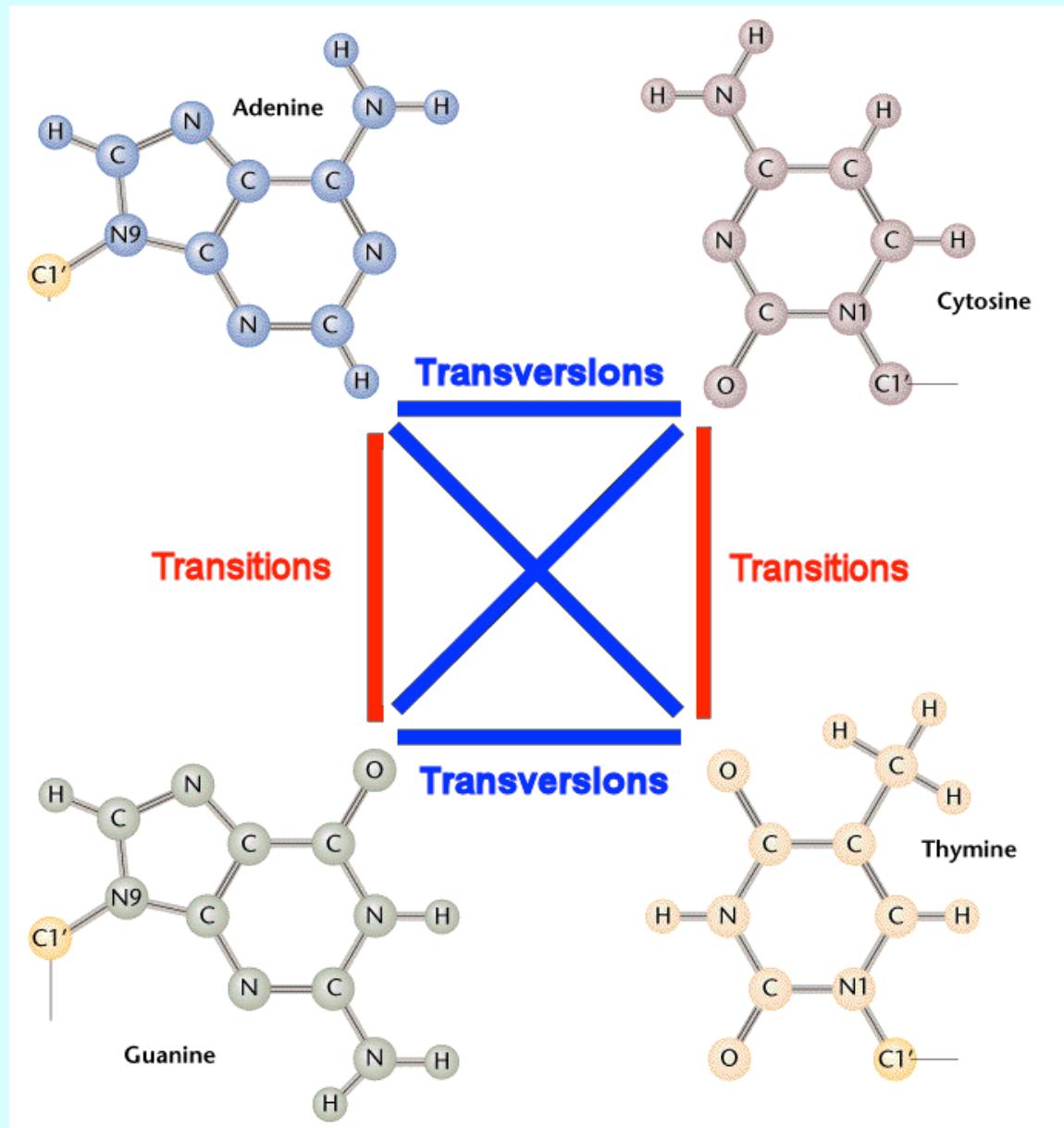
Origem: oxidação intracelular, radiações ionizantes

Efeito: alterações várias da estrutura do DNA

# Tipos de mutação

Transições: mesmo tipo de base

Transversões: tipo de base diferente

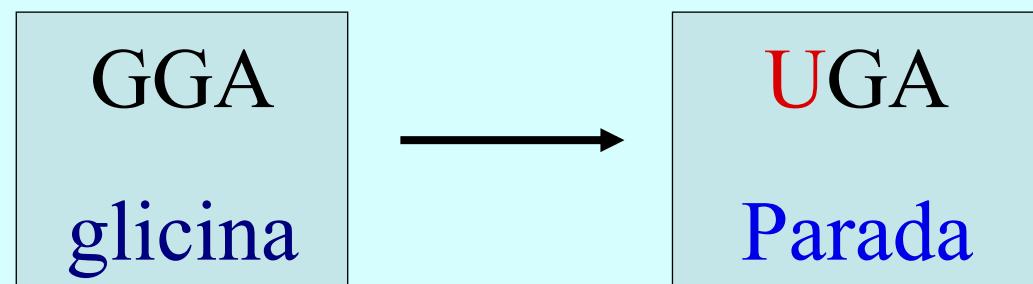
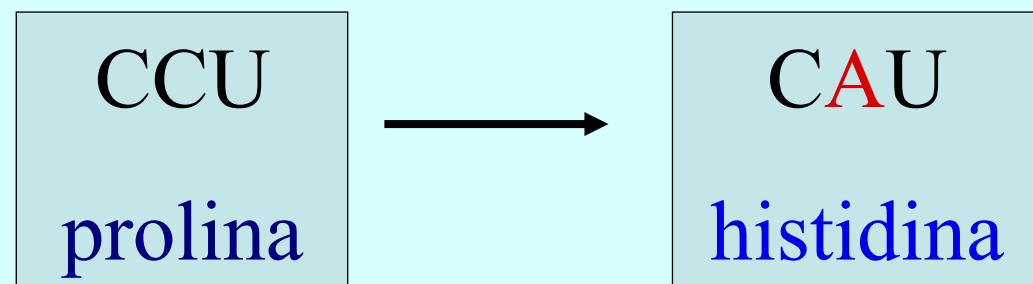
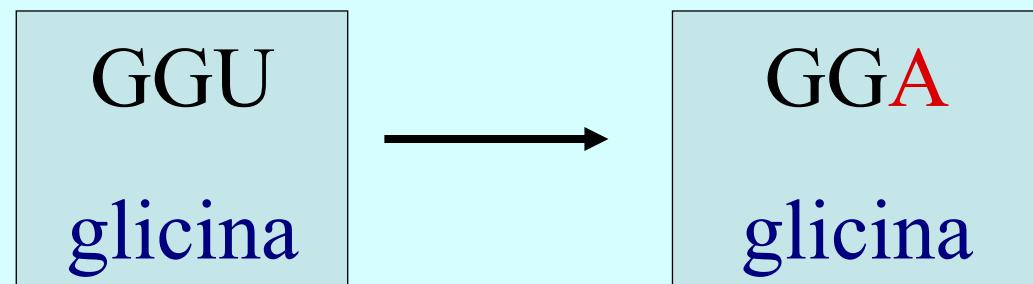


# Tipos de mutação (quanto ao efeito)

Sinônimas: mesmo aminoácido  
**(sem troca de sentido)**

Não sinônimas:  
aminoácido diferente  
**(com troca de sentido,  
missense)**

Sem sentido:  
aminoácido -> parada  
("nonsense")

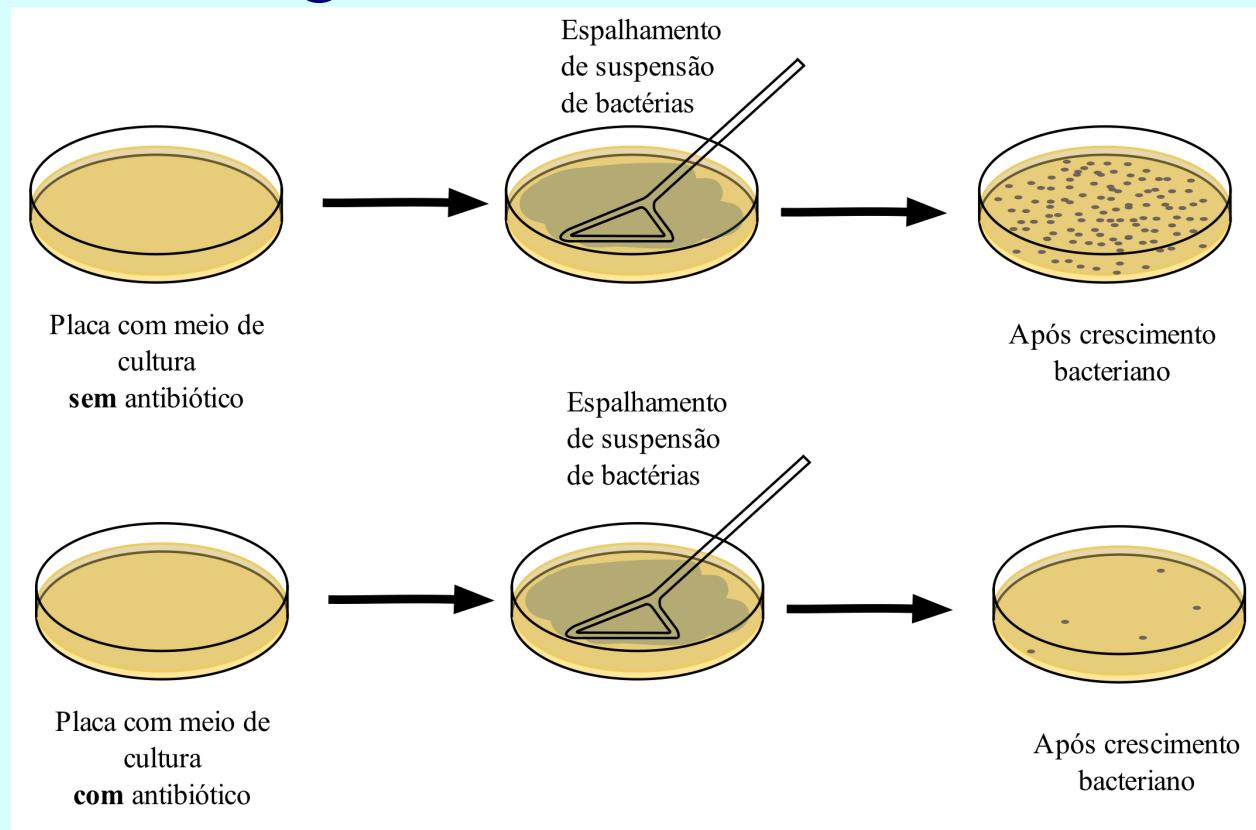


# Mutações e evolução

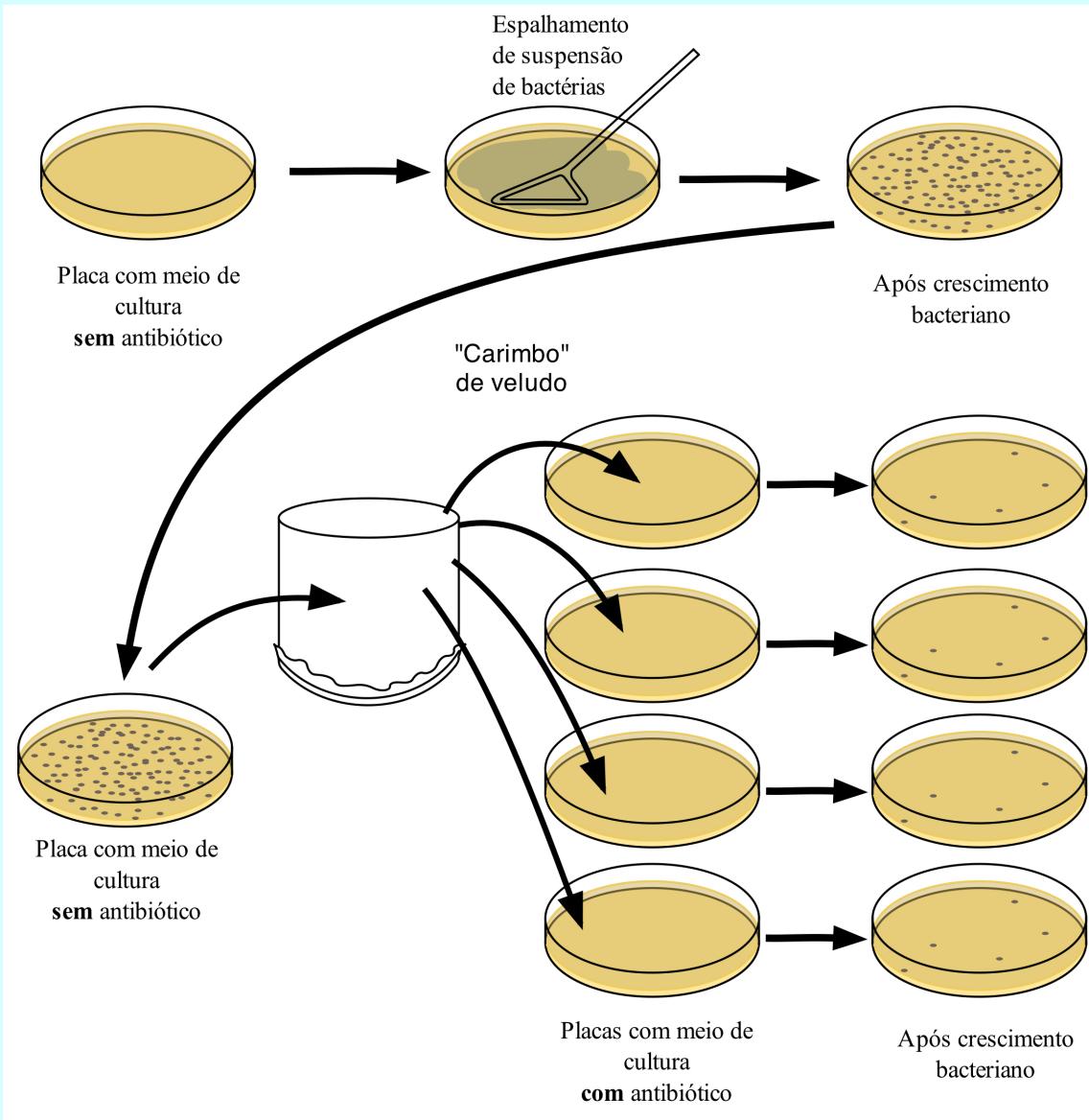
As mutações surgem por causa das mudanças ambientais ou ocorrem independentemente delas?

# Mutações e evolução

Experimento: Se colocarmos bactérias que são sensíveis a um determinado antibiótico em um meio de cultura com antibiótico, algumas colônias crescerão:



# Mutações e evolução



Experimento do  
“carimbo de  
veludo”  
(Lederberg e  
Lederberg, 1952)

# Alterações estruturais

